



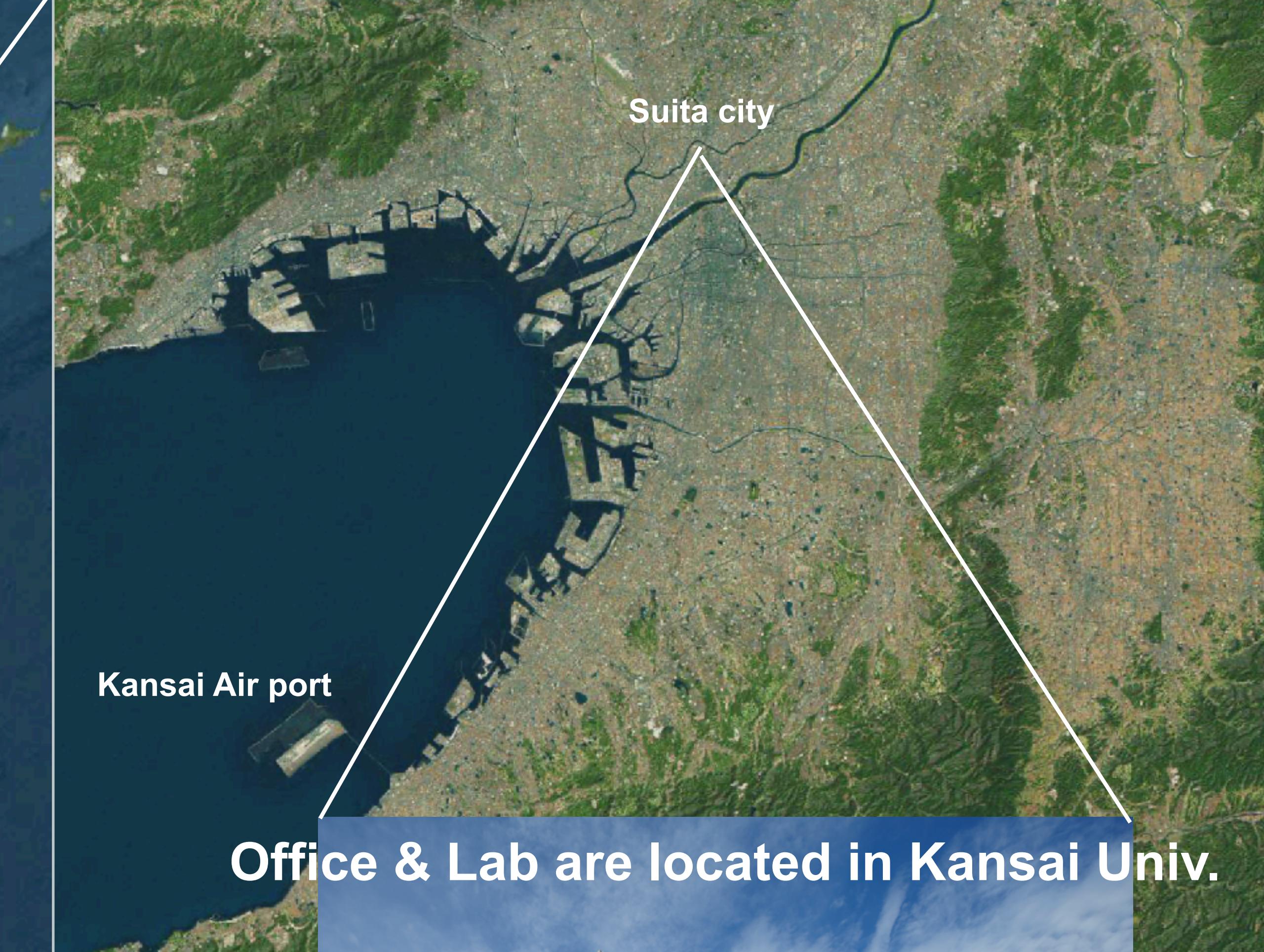
iELECTROLYTE

KANSAI JAPAN

Company Profile



- Founded: Apr. 2014
- Locating: Kansai Univ., Osaka, JAPAN



iElectrolyte accelerates to introduce low-environmental impact water-based positive electrode producing process from lab to market.

Business field



Material supply

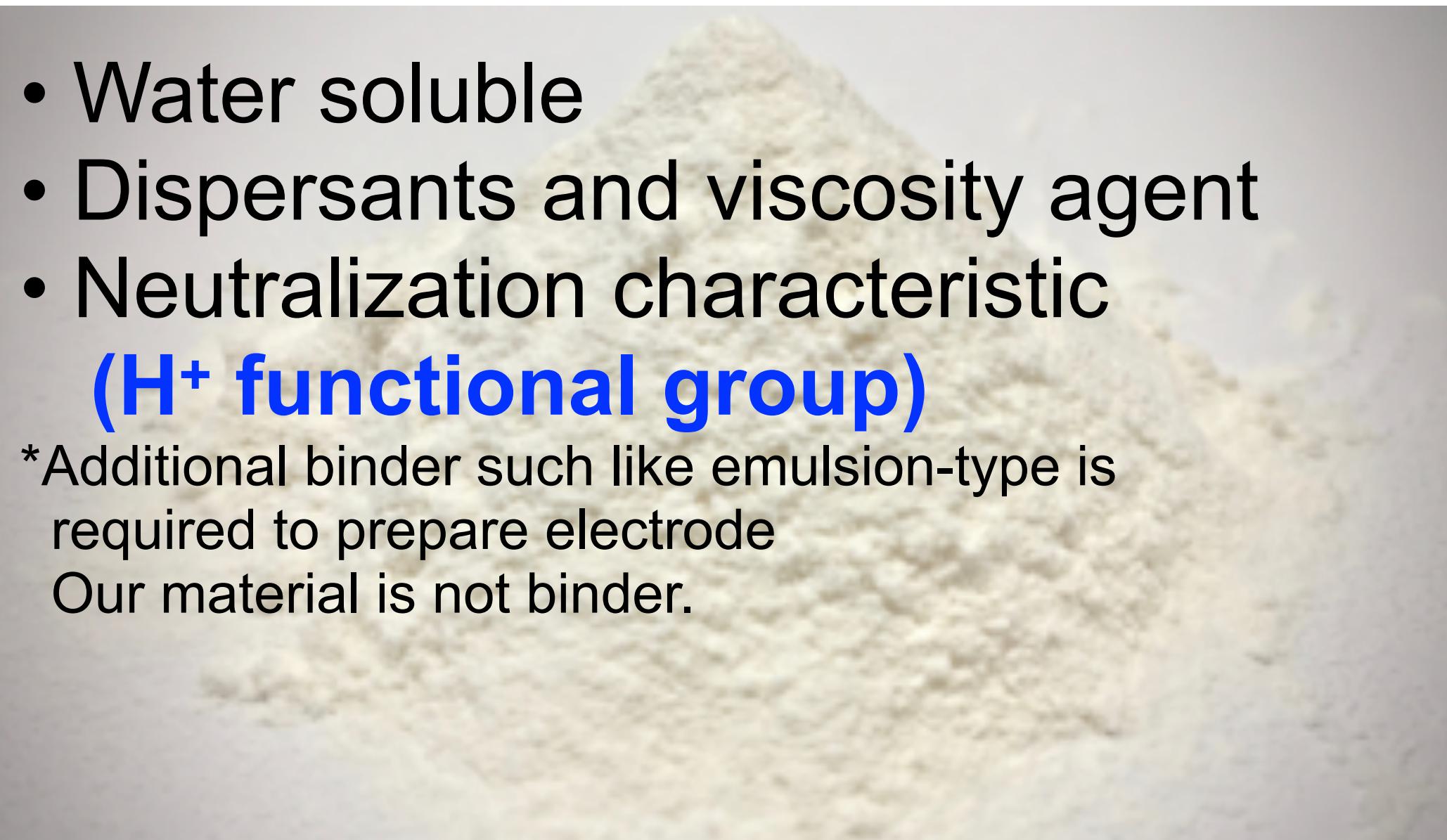
R&D support

- Process
- Material combination
- Cell test(Coin cell and Pouch cell)

License opportunity

Our material controls pH condition (pH <10) of water-based positive electrode slurry



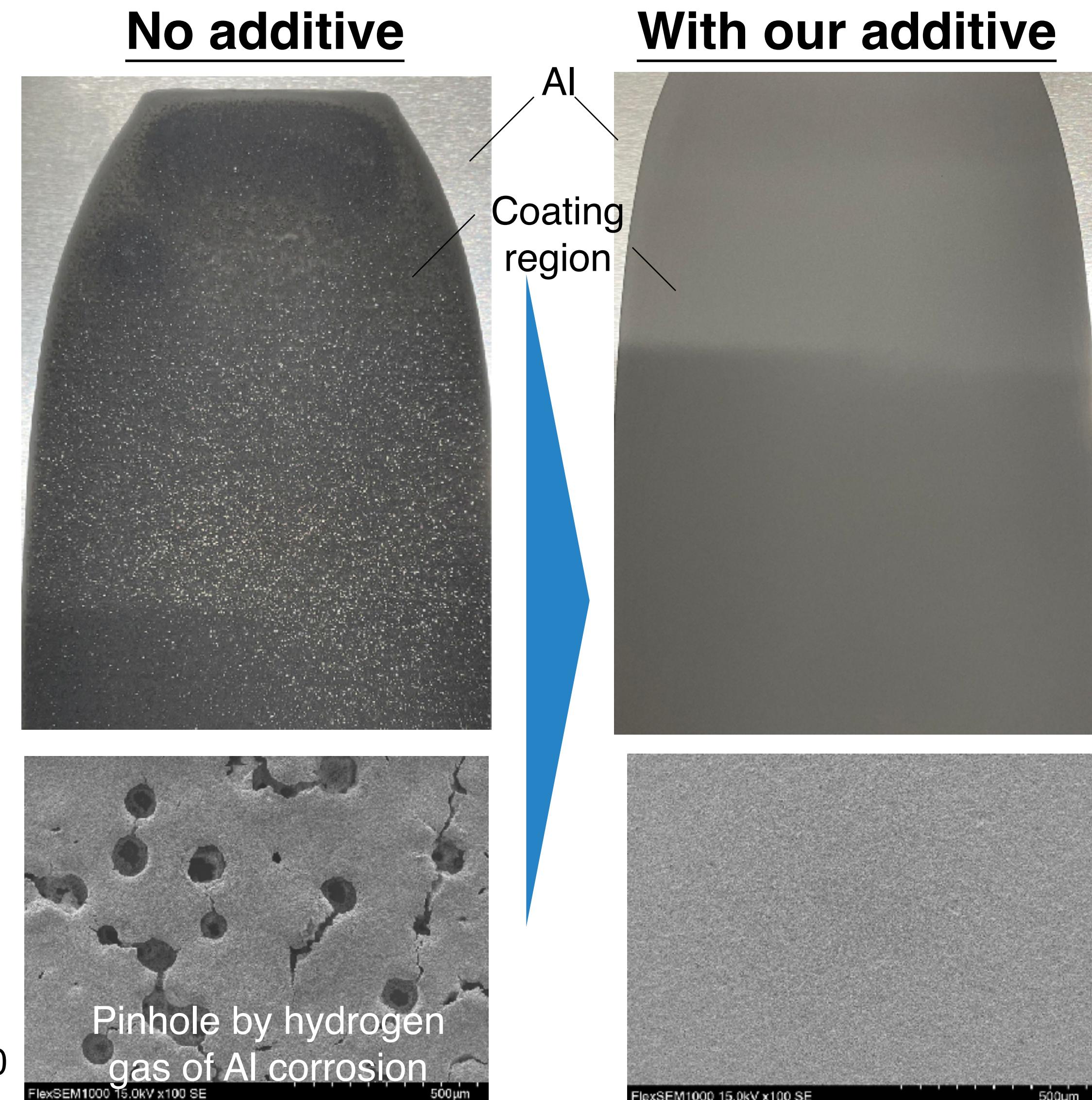
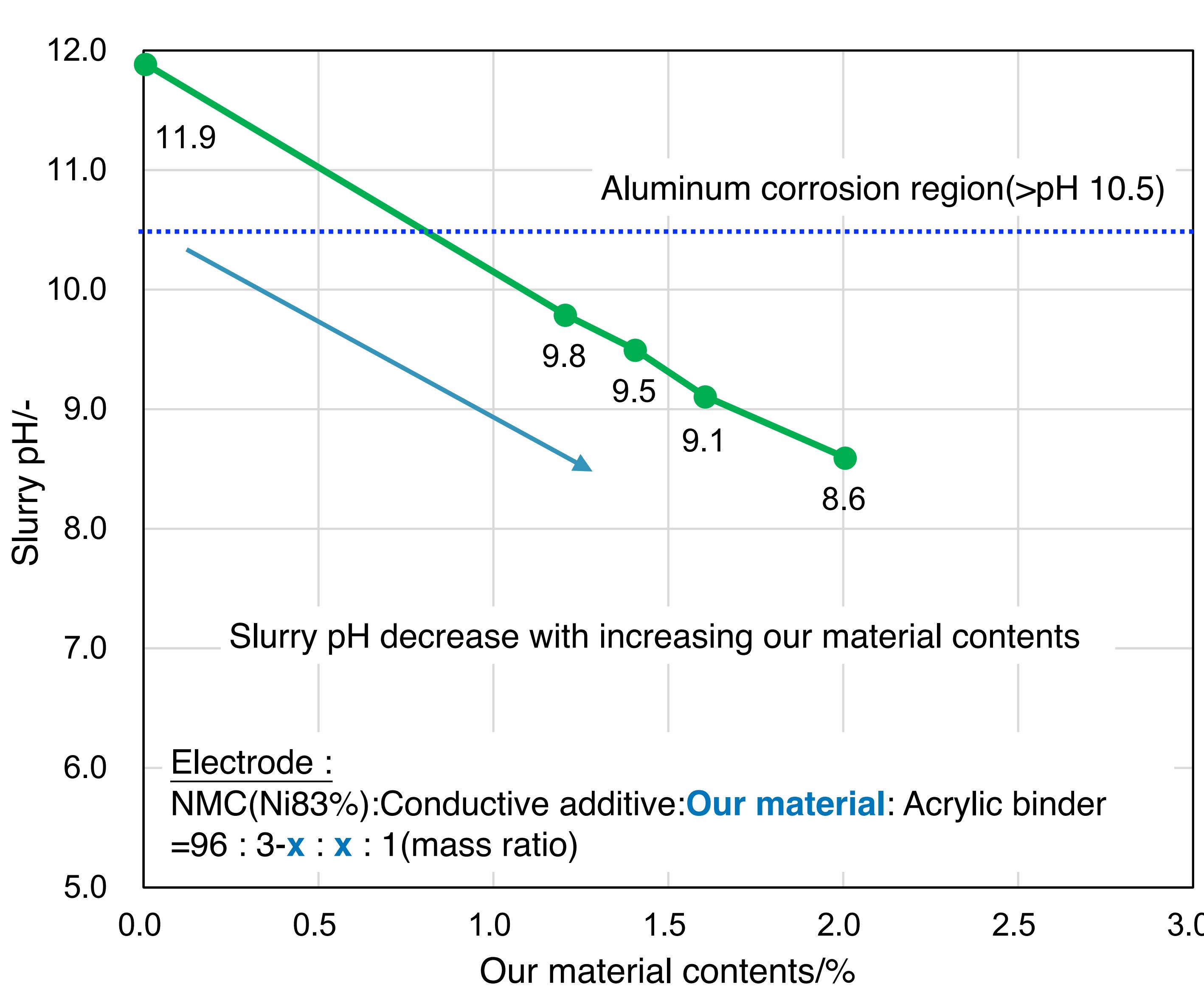
- Water soluble
 - Dispersants and viscosity agent
 - Neutralization characteristic
(H⁺ functional group)
- *Additional binder such like emulsion-type is required to prepare electrode
Our material is not binder.
- 

High alkaline condition(pH >10.5)
→Al current collector suffers corrosion

→ Our material + OH⁻ → H₂O
(H⁺ functional group)

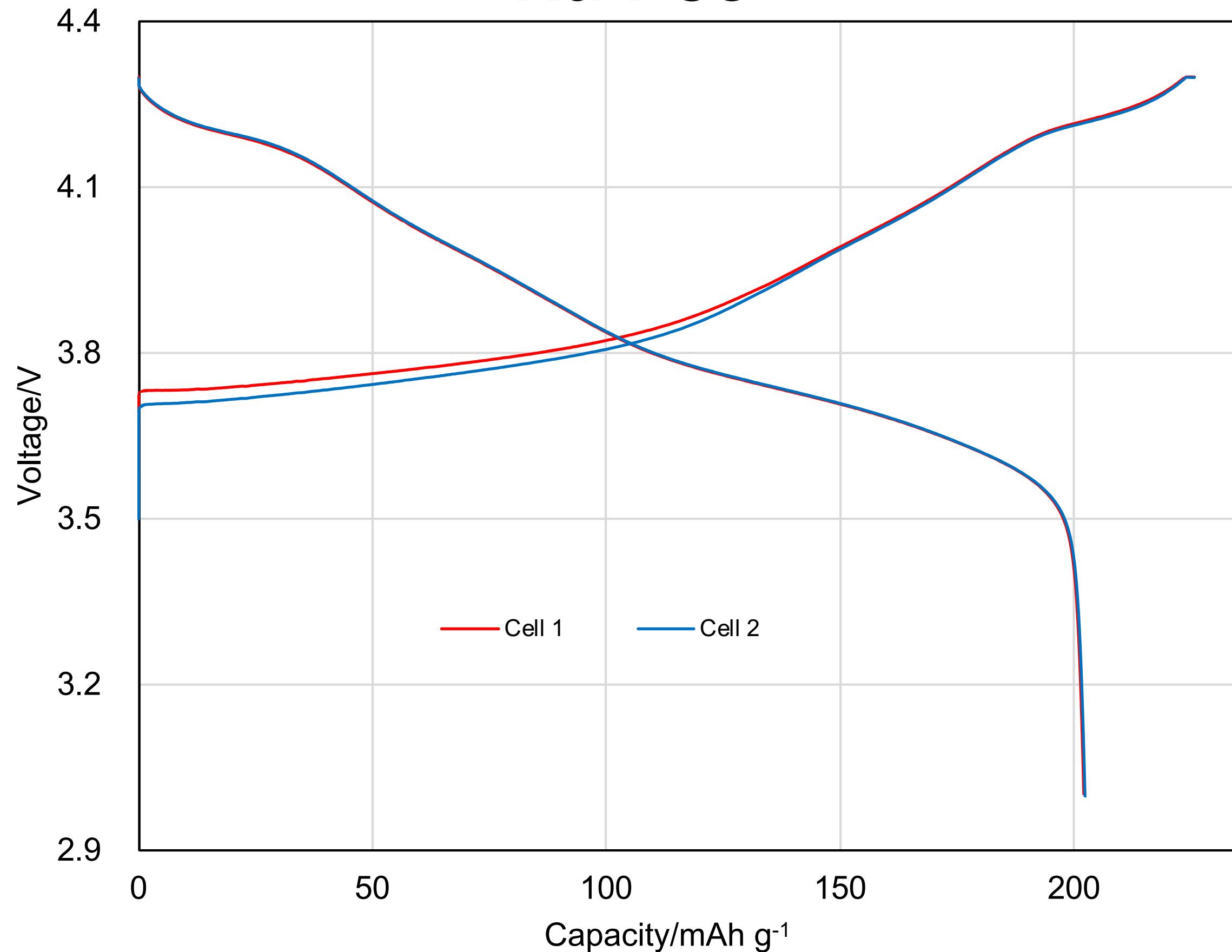
Reducing alkaline content(pH <10)
→No Al current collector corrosion

Impact of our material to aqueous positive electrode slurry pH behavior

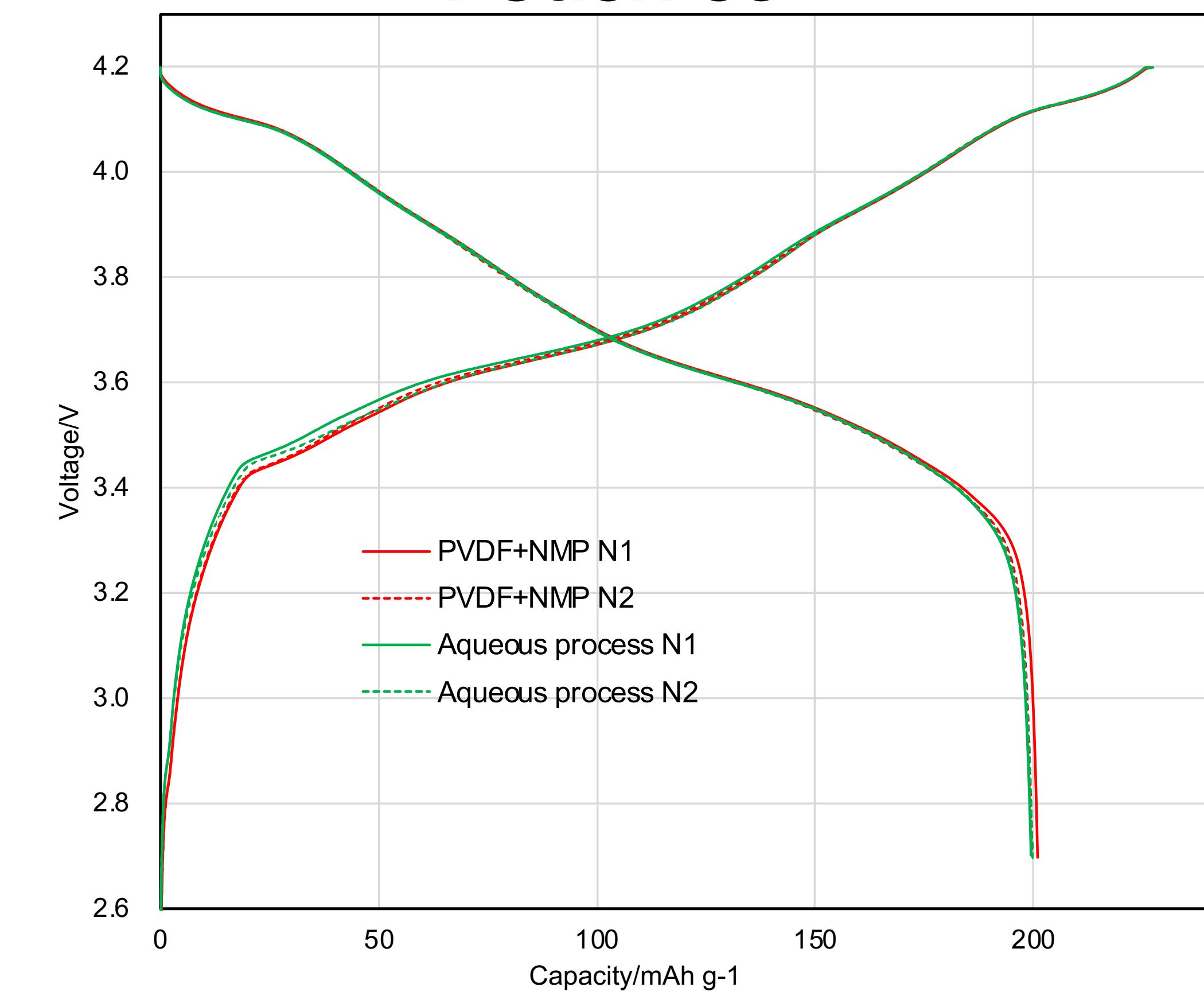


Electrochemical characteristic

Half-cell



Pouch cell



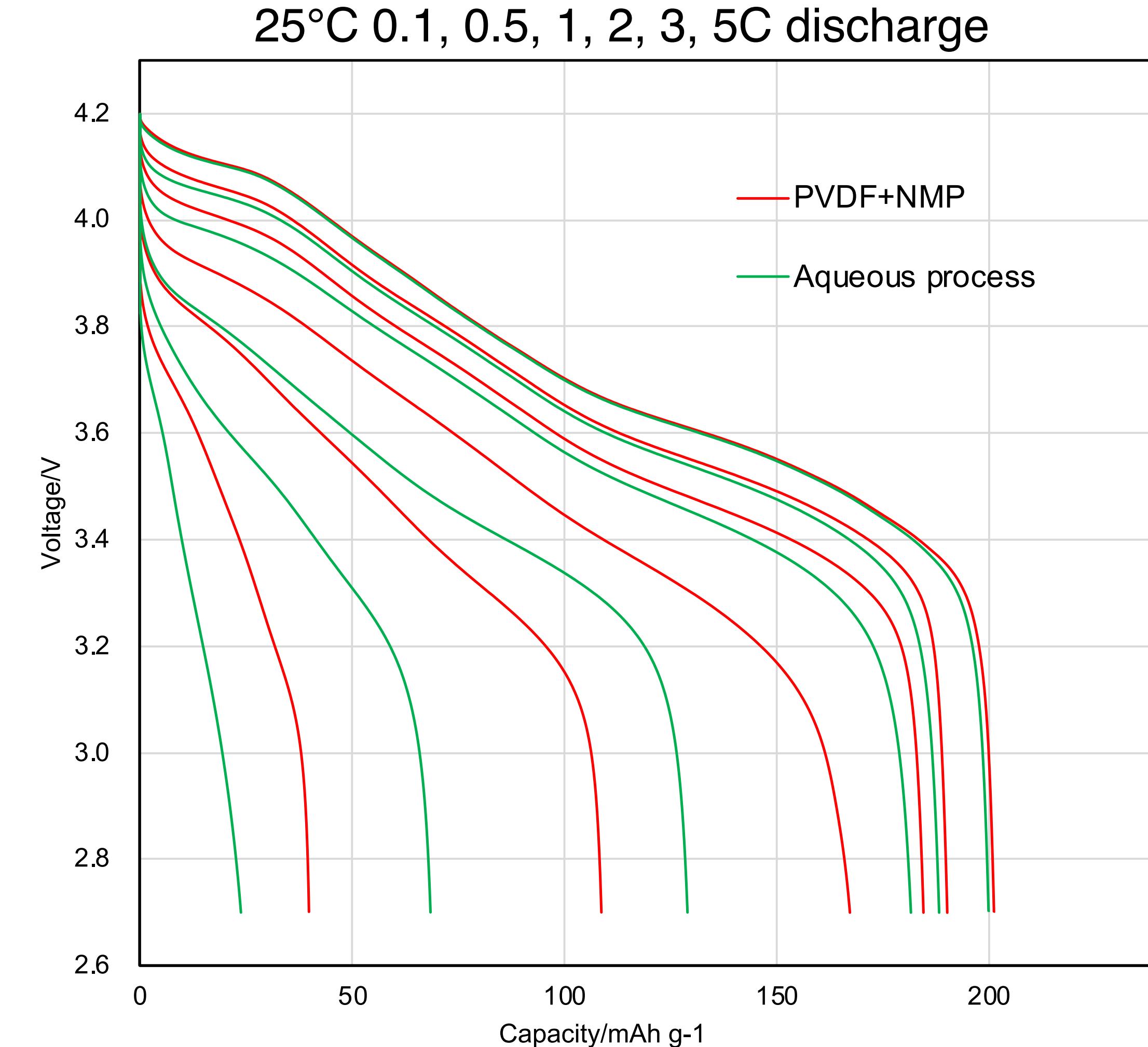
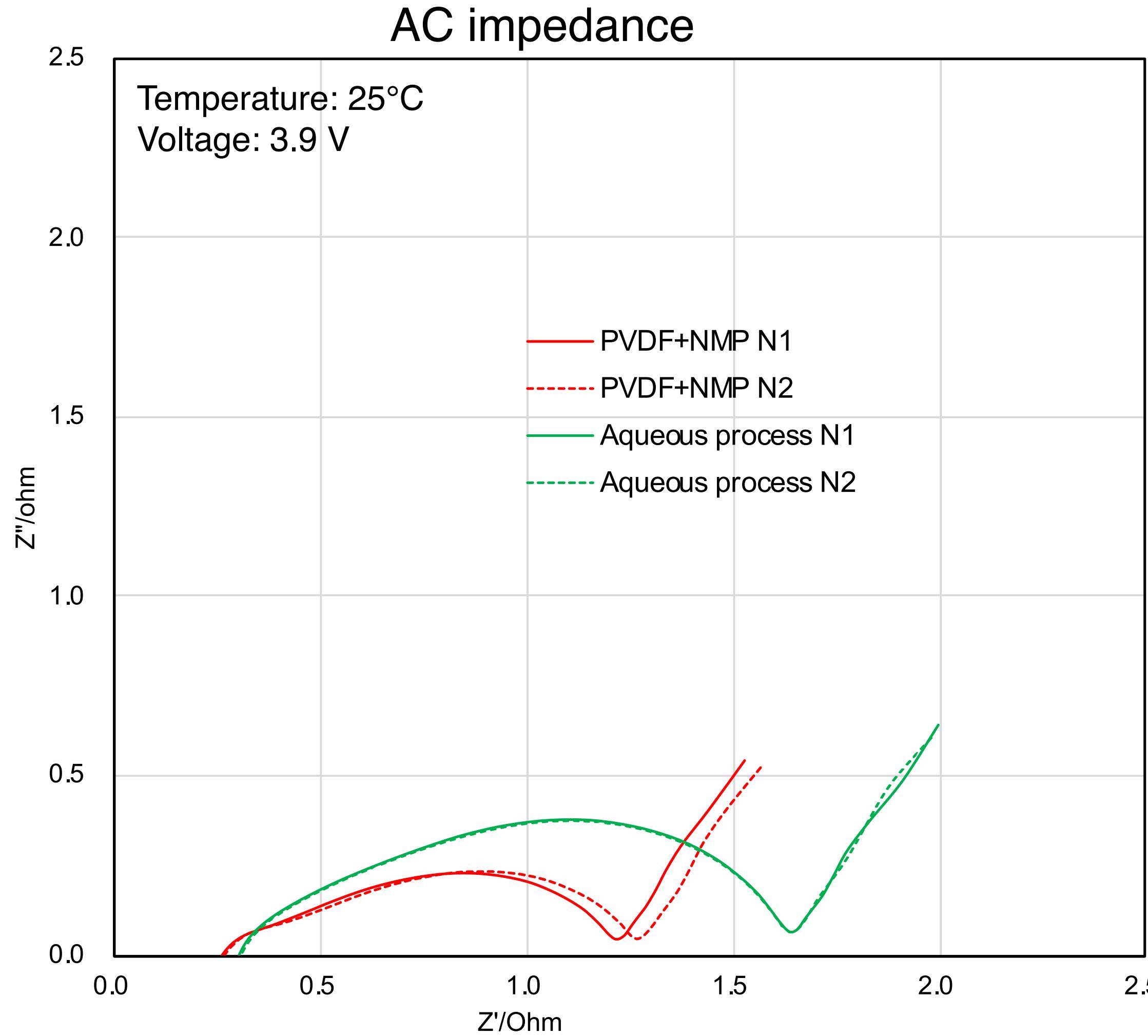
Electrode formulation

NMC(Ni83%): C65: Our material: Acrylic binder=94.5: 2.0: 2.0: 1.5
CE: Lithium foil(13φ)
Electrolyte: 1M LiPF₆/EC:DMC(1:1)
Cell: Coin cell

Cathode :

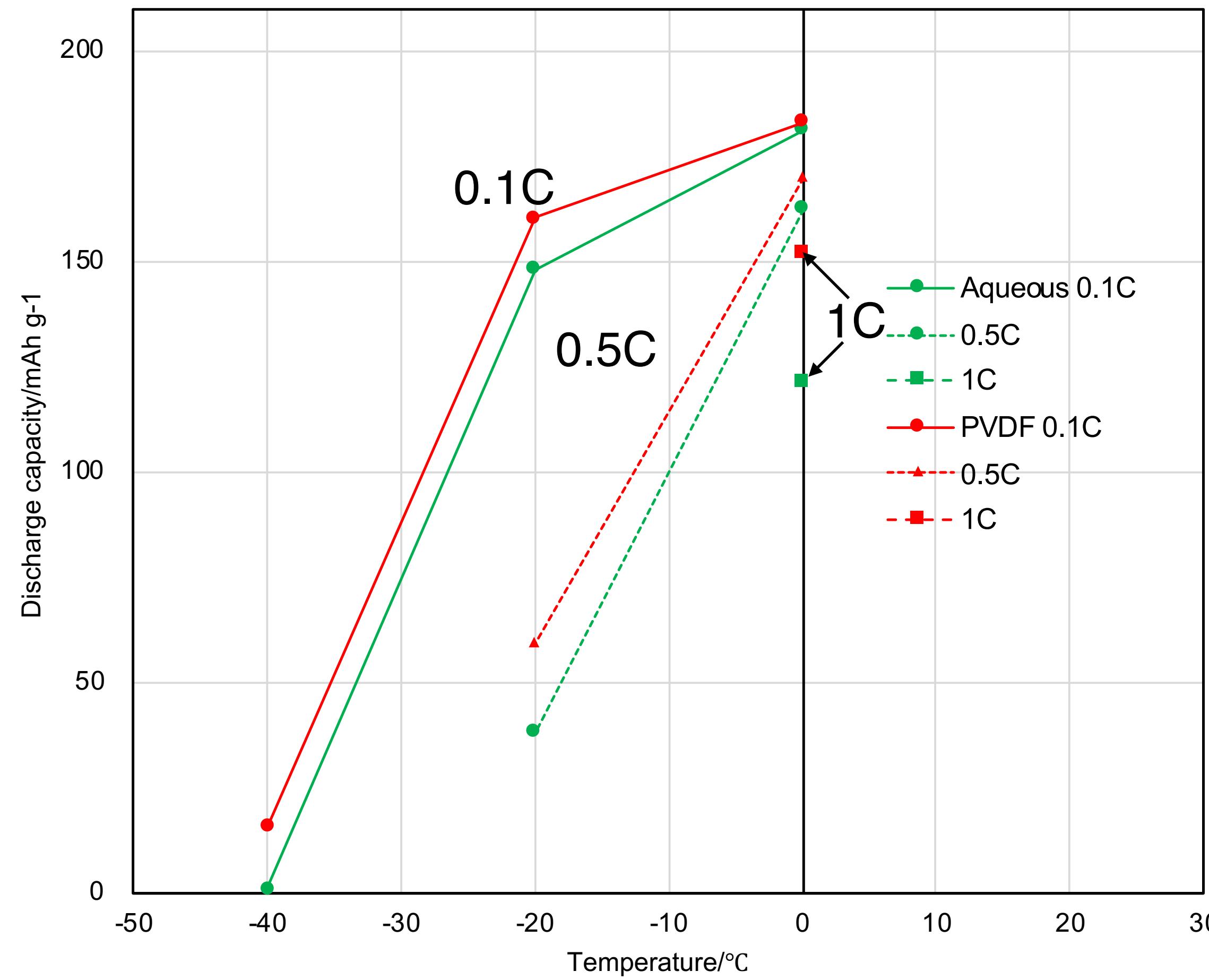
NMC(Ni83%): C65: PVDF = 96.0 : 2.0 : 2.0 (20 mg/cm²)
NMC(Ni83%): C65: Our material: Acrylic binder=94.0 : 2.0 : 2.0 : 2.0(20 mg/cm²)
Anode : Artificial graphite
Electrolyte : 1M LiPF₆/EC:DEC(3:7)+2%VC
Cell : Single layer pouch cell 55~60 mAh

Electrochemical characteristic

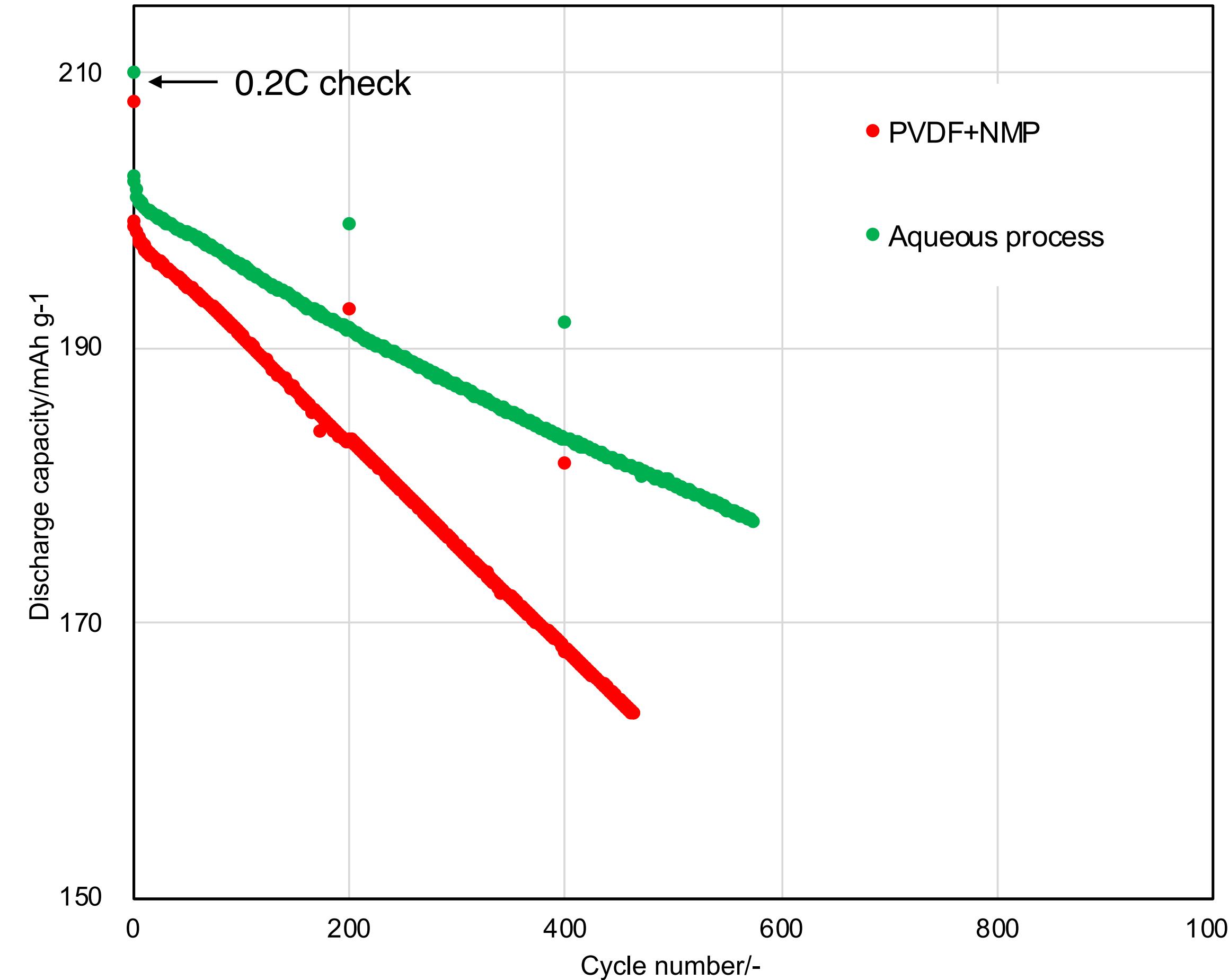


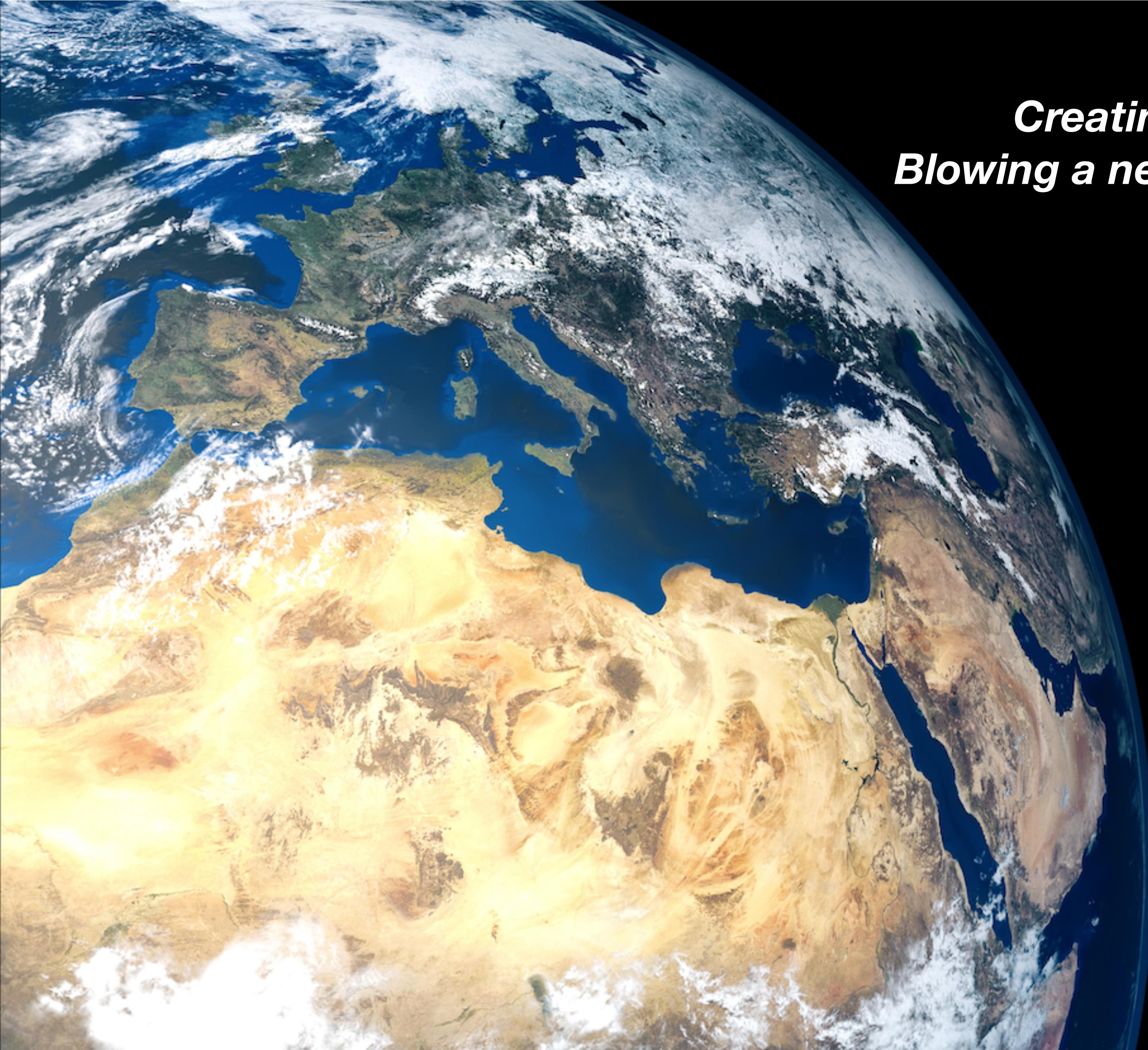
Electrochemical characteristic

0, -20, -40°C discharge



60°C 1C/1C cycle(4.2-2.7V)





*Creating and growing technology
Blowing a new wind toward electrochemistry*



Contact
Takuya Takahashi
Principal researcher
takahashi@ielectrolyte.net
+81-6-6318-5717